2013 Vol. 15, No. 11 2746–2749

Electroreductive Intermolecular Coupling of 3-Methoxycarbonylindoles with Ketones

Naoki Kise,* Akinori Sueyoshi, Shin-ichirou Takeuchi, and Toshihiko Sakurai

Department of Chemistry and Biotechnology, Graduate School of Engineering, Tottori University, 4-101, Koyama-cho Minami, Tottori 680-8552, Japan

kise@bio.tottori-u.ac.jp

Received April 19, 2013

ABSTRACT

The electroreductive coupling of 1-alkoxycarbonyl-3-methoxycarbonylindoles with aromatic ketones in the presence of chlorotrimethylsilane gave *cis*-adducts stereoselectively. The *cis*-adducts were readily transformed to *trans*-adducts by treatment with catalyst DBU. On the other hand, the electroreductive coupling of 1-methyl-3-methoxycarbonylindole with aliphatic ketones in isopropanol afforded *trans*-adducts exclusively. The adducts are the precursors for the synthesis of 2-substituted 3-methoxycarbonylindoles and indolines.

To date, a variety of reductive cross-couplings of unsaturated groups with ketones have been developed using SmI₂ as a reducing agent, and they have been exploited for the synthesis of a number of natural products and physiologically active substances.¹ Recently, Reissig and co-workers have reported the SmI₂-promoted reductive intra-² and intermolecular³ couplings of indole derivatives with ketones to synthesize indolidine heterocycles stereoselectively. On the other hand, we disclosed that the reductive intramolecular coupling of indole derivatives

with ketones was also realized by electroreduction in isopropanol⁴ and *trans*-cyclized products were stereoselectively formed similarly to the SmI₂-promoted cyclization^{2a} (Scheme 1). In this context, we report herein the electroreductive intermolecular coupling of *N*-substituted 3-methoxycarbonylindoles with aromatic and aliphatic ketones, since the coupled products are expected to be useful precursors for the synthesis of 2-substituted 3-methoxycarbonylindoles⁵ and indolines. The electroreduction of 1-alkoxycarbonyl-3-methoxycarbonylindoles with aromatic ketones in the presence of chlorotrimethyl-silane (TMSCl)⁶ gave intramolecularly coupled products

⁽¹⁾ For recent reviews on the reductive coupling with SmI₂: (a) Kagan, H. B. *Tetrahedron* **2003**, *59*, 10351. (b) Berndt, M.; Gross, S.; Hoelemann, A.; Reissig, H.-U. *Synlett* **2004**, *422*. (c) Edmonds, D.; Johnston, D.; Procter, D. J. *Chem. Rev.* **2004**, *104*, 3371. (d) Gopalaiah, K.; Kagan, H. B. *New J. Chem.* **2008**, *32*, 607. (e) Rudkin, I. M.; Miller, L. C.; Procter, D. J. *Organomet. Chem.* **2008**, *34*, 19. (f) Nicolaou, K. C.; Ellery, S. P.; Chen, J. S. *Angew. Chem., Int. Ed.* **2009**, *48*, 7140. (g) Beemelmanns, C.; Reissig, H.-U. *Chem. Soc. Rev.* **2011**, *40*, 2199.

^{(2) (}a) Gross, S.; Reissig, H.-U. *Org. Lett.* **2003**, *5*, 4305. (b) Blot, V.; Reissig, H.-U. *Eur. J. Org. Chem.* **2006**, 4989. (c) Beemelmanns, C.; Blot, V.; Cross, S.; Lentz, D.; Reissig, H.-U. *Eur. J. Org. Chem.* **2010**, 2716. (d) Beemelmanns, C.; Reissig, H.-U. *Amgew. Chem., Int. Ed.* **2010**, *49*, 8021. (e) Beemelmanns, C.; Lentz, D.; Reissig, H.-U. *Chem.—Eur. J.* **2011**, *17*, 9720.

⁽³⁾ Blot, V.; Reissig, H.-U. Synlett 2006, 2763.

⁽⁴⁾ Kise, N.; Mano, T.; Sakurai, T. Org. Lett. 2008, 10, 4617.

⁽⁵⁾ For recent reports on the synthesis of 2-substituted 3-alkoxycarbonylindoles, see: (a) Cui, S.-L.; Wang, J.; Wang, Y.-G. J. Am. Chem. Soc. 2008, 130, 13526. (b) Zhou, L.; Doyle, M. P. J. Org. Chem. 2009, 74, 9222. (c) Huestis, M. P.; Chan, L.; Stuart, D. R.; Fagnou, K. Angew. Chem., Int. Ed. 2011, 50, 1338. (d) Neumann, J. J.; Rakshit, S.; Dröge, T.; Würtz, S.; Glorius, F. Chem.—Eur. J. 2011, 17, 7298. (e) He, Z.; Liu, W.; Li, Z. Chem.—Asian. J. 2011, 6, 1340. (f) Bunescu, A.; Wang, Q.; Zhu, J. Synthesis 2012, 44, 3811. (g) Nguyen, H. H.; Kurth, M. J. Org. Lett. 2013, 15, 364.

⁽⁶⁾ For our recent reports on the electroreductive coupling in the presence of TMSCl, see: (a) Kise, N.; Isemoto, S.; Sakurai, T. *Org. Lett.* **2009**, *11*, 4902. (b) Kise, N.; Sakurai, T. *Tetrahedron Lett.* **2010**, *51*, 70. (c) Kise, N.; Isemoto, S.; Sakurai, T. *J. Org. Chem.* **2011**, *76*, 9856. (d) Kise, N.; Isemoto, S.; Sakurai, T. *Tetrahedron* **2012**, *68*, 8805.

with *cis*-stereoselectivity. The *cis*-adducts could be completely isomerized to the *trans*-adducts by treatment with catalyst (cat.) DBU. In contrast, the electroreductive coupling of 1-methyl-3-methoxycarbonylindole with aliphatic ketones in isopropanol using an undivided cell afforded *trans*-adducts exclusively. We investigated the reaction mechanisms of these reactions and found that the mechanisms of the electroreductive couplings with aromatic and aliphatic ketones are different.

Scheme 1. Electroreductive Intramolecular Coupling of Indole Derivatives with Ketones

First, we attempted the electroreductive coupling of 3-methoxycarbonylindoles 1a-d with benzophenone (2a) as a representative of aromatic ketones (Scheme 2). Unfortunately, 1a-d were recovered and benzhydrol was produced from 2a by the electroreduction in isopropanol.⁴ Therefore, we explored other conditions for the electroreductive coupling and found that the electroreduction of 1-alkoxycarbonyl-3-methoxycarbonylindoles 1a and 1b with 2a (2 equiv) in the presence of TMSCl (5 equiv) in THF solvent gave coupled products 3a and 3b in 88% and 72% yields, respectively (Scheme 2, runs 1 and 2). These products were formed with high stereoselectivity, and the major isomer of 3a was determined to be cis by X-ray crystallography (Supporting Information). The cis-isomers of 3a and 3b could be completely converted to the corresponding trans-isomers by treatment with a catalytic amount of DBU in THF at room temperature. Desilylation of the *trans*-isomers of **3a** and **3b** by treatment with 0.1 M HCl in methanol at 0 °C gave alcohols trans-4a and trans-4b. The relative configuration of trans-4b was confirmed by X-ray crystallography. When the electroreductive coupling of 1-methyl-3-methoxycarbonylindoles (1c) with 2a was carried out under the same conditions, no coupled product was obtained and trimethylsilyl ether of benzhydrol was produced. In place of 2a, 4,4'-difluoroand 4,4'-dimethoxybenzophenones (2b and 2c) were employed as aromatic ketones, and the results were summarized in Scheme 2 (runs 3–6). Although the cis-selectivity in the reductive coupling of 2b was somewhat lower (runs 3 and 4) than in that of 2a, similar yields of the coupled products 3c-f were obtained. Isomerization of the cisisomers of 3c-f with cat. DBU gave *trans*-3c-f exclusively.

Next, the electroreduction of **1a** with phenyl alkyl ketones **2d**—**f** was carried out under the same conditions as above (Scheme 3). Since coupled products were formed as complex mixtures of diastereomers, the crude products were treated with cat. DBU in THF and 0.1 M HCl in methanol successively to facilitate product isolation. All products **4g**—**i** seemed to be obtained as two diastereomers of *trans*-isomers. In the reaction with acetophenone (**2d**),

the yield of 4g was low (30%) owing to the formation of pinacols from 2d and unreacted 1a was recovered (>60%). Bulkier ketones 2e and 2f brought about 4h (48%) and 4i (54%) in somewhat higher yields.

Scheme 2. Electroreductive Coupling of 1-Alkoxycarbonyl-3-methoxycarbonylindoles with Diaryl Ketones^a

run	indole	ketone	% yield ^a (dr)			
			cis-3	trans-3	trans-4	
1	1a	2a	3a 88 (95:5)	3a quant	4a 95	
2	1b	2a	3b 72 (100:0)	3b quant	4b 90	
3	1a	2b	3c 77 (67:33)	3c quant	4c 93	
4	1b	2b	3d 81 (92:8)	3d quant	4d 91	
5	1a	2c	3e 73 (88:12)	3e quant	4e 92	
6	1b	2c	3f 68 (100:0)	3f quant	4f 93	

^a Isolated yield.

Scheme 3. Electroreductive Coupling of 1,3-Dimethoxycarbonyllindole with Phenyl Alkyl Ketones

In addition, we attempted the electroreductive coupling of 1-alkoxycarbonyl-3-methoxycarbonylindoles 1a and 1b with acetone (5a) as an aliphatic ketone. However, no coupled product was obtained and hydrogenated indolines were formed from 1a,b. After the survey of substrates and conditions for the electroreductive coupling with 5a, we found that 1-methyl-3-methoxycarbonylindole (1c) gave the expected product 6a in 73% yield as a sole stereoisomer together with a small amount of hydrogenated indoline 7 (10%) by the electroreduction with 5a (5 equiv) in isopropanol using an undivided cell⁴ (Scheme 4). The stereoconfiguration of the obtained 6a seemed to be trans, since NOE could not be observed between 2-H and 3-H protons in the ¹H NMR analysis of **6a**. Other aliphatic cyclic ketones 5b and 5c also afforded the coupled products 6b (60%) and 6c (58%), respectively. The relative configuration of **6b** was confirmed to be trans by X-ray crystallographic analysis.

Org. Lett., Vol. 15, No. 11, 2013

Scheme 4. Electroreductive Coupling of 1-Methyl-3-methoxy-carbonylindoles with Aliphatic Ketones^a

CO₂Me

+ R¹ R² + 2e

i-PrOH

1c 5a-c

$$CO_2Me$$
 R^1
 R^2
 R^1
 R^2
 R^1
 R^2
 R^2
 R^1
 R^2
 R^2

run	ketone	R ¹	R ²	% yielda		
Tun				trar	rs-6	7
1	5a	CH ₃	CH ₃	6a	73	10
2	5b	-(CH ₂) ₄ -		6b	60	24
3	5c	-(CH ₂) ₅ -		6c	58	27

^a Isolated yield

The cyclic voltamograms of aromatic ketones 2a and 2d in 0.03 M Bu₄NClO₄/DMF on a platinum cathode showed a first reduction peak at -1.87 and -2.14 V vs SCE, respectively, while those of 1-alkoxycarbonyl 3-methoxycarbonylindoles 1a and 1b under the same conditions revealed no reduction peak from 0 to -2.50 V vs SCE. These results suggest that 2a and 2d are more reducible than 1a and 1b. Therefore, the electroreductive coupling of 1a with 2a was supposed to be initiated by the reduction of 2a and the reaction mechanism can be presumed as illustrated in Scheme 5. Carbanion A is formed by the two-electron transfer to 2a and O-silylation with TMSCl. Since the electroreduction of 1c with 2a gave no coupled product as described above, it is likely that the active species is not an O-silylated radical but anion A. The nucleophilic addition of A to the 2-position of 1a and subsequent O-silylation of the resulting enolate anion give silvl ketene acetal **B**. The labile B is readily desilylated to ester 3a during workup. In this desilylation, protonation at the 3-position of **B** occurs from the less bulky side, that is the opposite side of the substituent at the 2-position, to produce *cis-3a* stereoselectively. The kinetically controlled product cis-3a was completely transformed to trans-3a by treatment with DBU. This result indicates that trans-3a is thermodynamically more stable than cis-3a. The DFT calculations (Supporting Information) also support this presumption, since *trans-3a* is much more stable (8.39 kcal/mol in THF) than cis-3a.

On the other hand, the electroreductive intermolecular coupling of **1c** with aliphatic ketone **5a** is presumed to be promoted by the reduction of **1c** as the already reported intramolecular coupling, ⁴ in which one-electron transfer to the 3-methoxycarbonyl moiety in the substrate takes place (Scheme 6). DFT calculations of radical anion **C** generated by the one-electron transfer to **1c** show the highest spin density exists at the C2 carbon and a negative charge is delocalized at the oxygen atoms. The radical anion **C**

attacks **5a** to give intermediate **D**, and then further reduction of **D** followed by protonation produces **6a**. In the protonation of the enolate anion in **D**, *cis*-**6a** may be produced preferentially. However, it is likely that *cis*-**6a** readily isomerized to thermodynamically more stable *trans*-**6a** under the basic conditions near the cathode. The DFT calculations suggest this assumption: *trans*-**6a** is lower in energy (3.46 kcal/mol in isopropanol) than *cis*-**6a**.

Scheme 5. Presumed Reaction Mechanism of Electroreductive coupling of 1,3-Di(methoxycarbonyl)indole with Benzophenone

Scheme 6. Presumed Reaction Mechanism of Electroreductive Coupling of 1-Methyl-3-methoxycarbonylindole with Acetone

$$\begin{array}{c} \text{CO}_2\text{Me} \\ \text{+e} \\ \text{N}_{\text{Me}} \\ \text{1c} \\ \text{T}_{\text{N}} \\ \text{N}_{\text{Me}} \\ \text{N}_{\text{Me}} \\ \text{T}_{\text{N}} \\ \text{N}_{\text{Me}} \\$$

The obtained coupled products **4** and **6** are promising as the precursors for the synthesis of 2-substituted 3-methox-ycarbonylindoles and indolines. The preliminary results of our ongoing study are shown in Scheme 7. Dehydration of *trans*-**4a** by refluxing in toluene in the presence of cat. *p*-TsOH gave 2-diphenylmethylindole **8** (78%). Deprotection of **8** was effected by treatment with TBAF⁷ to give **9** (94%). The *N*-unsubstituted 2-diphenylmethylindole **9**

2748 Org. Lett., Vol. 15, No. 11, 2013

⁽⁷⁾ Jaxquemard, U.; Bénéteau, V.; Lefoix, M.; Routier, S.; Mérour, J.-Y.; Coudert, G. *Tetrahedron* **2004**, *60*, 10039.

Scheme 7. Transformations of trans-4a and trans-6a

Scheme 8. Transformation of trans-6a to 14

trans-6a
$$\xrightarrow{\text{MsCl}}$$
 $\xrightarrow{\text{Et}_3\text{N}}$ $\xrightarrow{\text{CO}_2\text{Me}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{CI}}$ $\xrightarrow{\text{Ne}}$ $\xrightarrow{\text{Ne}$

was also obtained from *trans-4b* by treatment with 1 M HCl in methanol at 25 °C, although the yield decreased

(56%). In contrast, dehydration of *trans*-6a under the same conditions produced 2-isopropenylindoline 10. Dehydrogenation of 10 with DDQ and the following hydrogenation of resulting 11 afforded 2-isopropylindole 12.8 Unexpectedly, chloride 13 prepared from *trans*-6a was rearranged to 1,2,3,4-tetrahydroquinoline 14 by treatment with *t*-BuOK in THF (Scheme 8). The ring expansion of 13 to 14 seems to be promoted by the formation of cyclopropane intermediate E and its ring opening due to the reaction with water through iminium ion F. Although the stereoconfiguration was not determined, 14 was formed as a single stereoisomer according to ¹H and ¹³C NMR analyses. Further transformations of the adducts 4 and 6 to other indoles and indolines are in progress.

In conclusion, the electroreduction of 1-alkoxycarbonyl-3-methoxycarbonylindoles 1a and 1b with benzophenones 2a-c in the presence of TMSCl in THF gave intermolecularly coupled products 3a-f with cis-stereoselectivity. The cis-isomers of 3a-f were readily transformed to transisomers quantitatively by treatment with cat. DBU in THF. Desilylation of trans-3a-f afforded the corresponding alcohols *trans*-4a-f. The electroreduction of 1a and 1b with phenyl alkyl ketones 2d-f under the same conditions and subsequent isomerization followed by desilylation yielded *trans-4g-i* as mixtures of two diastereomers. The electroreduction of 1-methyl-3-methoxycarbonylindole (1c) with aliphatic ketones 5a-c in isopropanol gave intermolecularly coupled products *trans*-6a-c exclusively. The products 4 and 6 are the precursors for the synthesis of 2-substituted 3-methoxycarbonylindoles and indolines.

Supporting Information Available. Experimental procedures; characterization data for products; ¹H and ¹³C NMR spectra of products; X-ray crystallographic data (ortep) of *cis-3a*, *trans-4b*, and *trans-6b*; and the results of DFT calculations for **3a**, **6a**, and **C**. Crystallographic CIF files for *cis-3a*, *trans-4b*, and *trans-6b*. This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.

Org. Lett., Vol. 15, No. 11, 2013

⁽⁸⁾ Suresh, J. R.; Syam Kumar, U. K.; Ila, H.; Junjappa, H. *Tetra-hedron* **2001**, *57*, 781.